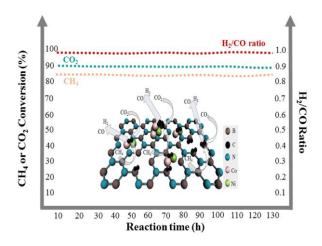
2D Boron Nitride Supported Nickel Catalyst for Dry Reforming of Methane

<u>Sanchai Kuboon</u>^a, Kajornsak Faungnawakij^a, Min Gao^b, Jiang Deng^c, Dengsong Zhang^c

- ^{a)} National Nanotechnology Center, National Science and Technology Development Agency
- b) Institute for Catalysis, Hokkaido University
- c) Department of Chemistry, Research Center of Nano Science and Technology, Shanghai University

2D materials such as boron nitride have been recently employed as promising nickel supports in dry reforming of methane owning to it high thermal telorance and inerness. Exfoliation of boron nitride could both yied thin layered boron nitride nanosheet and generate defect sites which is beneficial for occupying metal active sites and result in enhancement of catalytic performance. Good interaction between metal and support leads to well dispersed nickel over 2D boron nitride which could not only enhance CO₂ and CH₄ conversion but also reduce carbon formation. Systematic catalyst chracterization via physicochemical techniques and electron microscopes could provide strong evidence on this such interaciton. *In situ* diffuse reflectance infrared spectroscope as well as density-functional theory display insight information regarding dry reforming mechanism of methane over the 2D boron nitride supported nickel catalyst.

Corresponding author e-mail: sanchai@nanotec.or.th



Dry reforming of methane over nickel/boron nitride catalyst